High rate and selective etching of GaN, AlGa_xN, and AlN using an inductively coupled plasma

S. A. Smith,^{a)} C. A. Wolden, M. D. Bremser, A. D. Hanser, and R. F. Davis
*Department of Materials Science and Engineering, North Carolina State University, Raleigh,
North Carolina 27695-7919*

W. V. Lampert

Materials Directorate, Air Force Research Laboratory, Wright Patterson Air Force Base, Ohio 45433-7750

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The etching behavior of gallium nitride (GaN), aluminum gallium nitride (Al_xGa_{1-x}N), and aluminum nitride (AlN) has been systematically examined in an inductively coupled plasma (ICP) using Cl₂ and Ar as the reagents. Etch rates were strongly influenced by ICP power and dc bias, while relatively insensitive to pressure, flow rate, and gas composition. Maximum etch rates of 9800 Å/min for GaN, 9060 Å/min for Al_{0.28}Ga_{0.72}N, and 7490 Å/min for AlN were attained. The etch profiles were highly anisotropic over the range of conditions studied. The dc bias had to exceed certain voltages before significant etch rates were obtained. These values were < -20 V for GaN, -40 V for Al_{0.28}Ga_{0.72}N, and > -50 V for AlN. As such, increasing selectivity for GaN over Al_{0.28}Ga_{0.72}N and AlN was achieved at dc biases below -40 V. At -20 V, the GaN etch rates were 38 times greater than AlN and a factor of 10 greater than Al_{0.28}Ga_{0.72}N. These results demonstrate the importance of ion bombardment in the etching of these materials. © 1997 American Institute of Physics. [S0003-6951(97)03651-6]

The realized and future potential of GaN, AlN, and Al_xGa_{1-x}N for short wavelength light emitters and detectors, and high-temperature and high-power electronics is considerable.^{1,2} Etching anisotropic features is a crucial step in the fabrication of many of these devices. Wet chemical etching of GaN has to date produced only slow etch rates and isotropic etch profiles,³ both of which are undesirable for commercial applications. Dry etching is an attractive alternative from which one can achieve controlled degrees of anisotropy, high etch rates, material selectivity, low damage, and the ability to control an etch stop.

The four primary dry techniques that have been employed to etch GaN are reactive ion etching (RIE), electron cyclotron resonance etching (ECR), magnetron reactive ion etching (MIE), and inductively coupled plasma etching (ICP).⁴⁻⁶ The slowest etch rates and the lowest degree of anisotropy were determined for RIE. This has been attributed to the low plasma densities and the higher operating pressures inherent in this technique. The other three process routes are high density, low pressure alternatives to the parallel plate reactor. To date, the use of ECR has produced the fastest GaN etch rates of 13 000 Å/min with an ICI chemistry.⁷ Rates of 6875 Å/min were attained with an ICP system and a Cl₂/H₂/Ar mixture.⁸

Chlorine-based gases such as BCl₃, SiCl₄, and Cl₂ are the primary reagents that have been employed to etch the III-Nitrides. Vartuli *et al.* have reported a dry technique that etches GaN selectively with respect to AlN and Al_xGa_{1-x}N.^{9,10} Selectivities of 10 between GaN and AlN, and 4 between GaN and Al_{0.31}Ga_{0.69}N were reported. In this letter we report the results of a systematic study of ICP etching of III-Nitrides using a Cl₂/Ar chemistry. Selective etch-

ing of GaN relative to both AlN and Al_{0.28}Ga_{0.72}N was achieved by controlling the dc bias. The dependence of the etch rate on both ICP power and dc bias are also reported for the three materials. Lastly, the anisotropic nature of the etch profiles is demonstrated.

The ICP system was a custom built, 41 cm diameter by 58 cm tall, load-locked stainless-steel chamber. The rf power was coupled through a 32.4 cm diameter quartz window at the top of the chamber. The inductive source was a planar, 4 turn, 23 cm diameter copper coil which was connected to a rf power products 2 kW rf generator operating at 13.56 MHz via an autotuning matching network. Gas was fed into the chamber through a stainless-steel shower ring positioned level with the bottom of the quartz window. A water-cooled wafer chuck was mounted on a motor driven vertical translation stage which had 30.5 cm of travel. This allowed samples to be transferred between the load-lock chamber and the processing zone. A second 500 W rf source was connected to the wafer chuck to apply a controllable dc bias to the substrate. The substrate cooling water was maintained at 16 °C to prevent the baking of photoresist during etching. The chamber was evacuated by an Alcatel 900 ℓ/s turbomolecular pump which attained a base pressure of 10⁻⁷ Torr.

A magnetic bucket containing 240 Nd-Fe-B magnets was mounted on the outside perimeter of the chamber to increase the plasma density by confining the electrons to a central volume within the chamber. This reduced electron losses due to collisions with the chamber walls. The ions were also confined due to electrostatic coupling with the electrons. The former were not directly affected by the magnetic field.¹¹

The GaN, AlN, and Al_{0.28}Ga_{0.72}N samples used for this study were epitaxially grown on 6H-SiC-(0001) substrates via metalorganic vapor phase epitaxy (MOVPE) using trimethylaluminum (TMA) and triethylgallium (TEG) as the Al

^{a)}Permanent address: Materials Directorate, Air Force Research Laboratory, Wright Patterson Air Force Base, OH 45433-7750.

TABLE I. The experimental parameters examined, their ranges, and the base conditions employed in this study.

Parameter	Range	Base Cond.
ICP power	100–1100 W	500 W
DC bias	20–450 (–V)	–150 V
Pressure	1–9 mTorr	5 mTorr
Total flow rate	10–30 sccm	25 sccm
Cl ₂ Percentage	50–100%	80%
Ar percentage	0–50%	20%

and Ga sources, respectively, and NH₃ as the nitrogen source.^{12,13} An ≈ 100 nm AlN buffer layer was deposited on the SiC substrates prior to the growth of the GaN and the Al_{0.28}Ga_{0.72}N. Preparation of the samples for etching employed the sequence of applying a Ni coating, patterning with photoresist, and dipping into HNO₃ to etch away the Ni and into acetone to remove the photoresist. Just prior to entry into the etching system, the samples were dipped into 10% HF acid for 10 min to remove oxygen and carbon contaminants. Samples were attached to a 7.6 cm diameter anodized aluminum transport plate using vacuum grease which was mounted onto the wafer chuck. After entry into the system a base pressure of $\leq 5 \times 10^{-7}$ Torr was attained before the etching experiments were initiated.

The etch rates were determined by the step heights measured using a Dektak II profilometer. Anisotropy was determined by scanning electron microscopy (SEM). Other effects such as electrical damage, physical damage, and chemical contamination are currently being investigated.

The etching parameters studied and the ranges examined are summarized in Table I. The dc bias was varied instead of bias power since the former more directly determines the energy of the ions which strike the surface. The three samples were etched simultaneously to ensure accurate comparisons. It was determined that the gas phase concentration and total flow rate within the ranges shown in Table I had no significant impact on the etch rates ($< 20\%$). The etch rate was slightly more sensitive to pressure, varying 44% over the range investigated. A more extensive study of all parameters using optical emission spectrometry (OES) and mass spectrometry will be reported at a later date.

Figure 1 shows that the etch rates for the three materials as a function of ICP power for the three materials behaved similarly. The etch rates increased rapidly from 100 to 500 W. Above 500 W the rate of increase was slightly attenuated, although no maximum was reached. Between 900 and 1100 W, the etch rates of the Al_{0.28}Ga_{0.72}N closely matched those of the GaN. The observed increase in the etch rates with ICP power was likely due to increases in both the reactive chlorine concentration and the ion density. Additional diagnostics are necessary to distinguish between these two effects. The maximum etch rates achieved at an ICP power of 1100 W were 9140 Å/min for the GaN, 9060 Å/min for the Al_{0.28}Ga_{0.72}N, and 7490 Å/min for the AlN.

The etch profiles were highly anisotropic over all conditions examined. Figure 2 shows a typical profile obtained under the base conditions (see Table I) using nickel as the etch mask. The vertical striations were transferred from imperfections in the mask edge. The etched surface was smooth

FIG. 1. The etch rates of GaN, Al_{0.28}Ga_{0.72}N, and AlN as a function of ICP power.

and appeared to be free of physical damage such as pitting.

The dc bias had a very pronounced effect on the etch rates for all three materials, as shown in Fig. 3. Again, the behavior was qualitatively similar for the three materials. Firstly, each material had a threshold voltage below which the etch rates were very low. These voltages were: < -20 V for GaN, ~ -40 V for Al_{0.28}Ga_{0.72}N, and between -50 and -150 V for AlN. Above the threshold value the rates increased dramatically to -250 V at which point a plateau occurred for the three materials. Between -350 and -450 V the rates increased again, suggesting that an additional etching mechanism may be operative at these high biases. The maximum rates achieved at a dc bias of -450 V were 9800 Å/min for GaN, 8670 Å/min for Al_{0.28}Ga_{0.72}N, and 6700 Å/min for AlN.

The etch selectivity of GaN relative to AlN and Al_xGa_{1-x}N is of significant interest for the fabrication of Al_xGa_{1-x}N based heterostructure devices. One example is the etch penetration through a GaN capping layer to the Al_xGa_{1-x}N recessed gate in a high electron mobility transistor. Etching of the latter material should be minimal. Selective etching was achieved at low dc biases. The data from Fig. 3 were converted into selectivity and plotted in Fig. 4

FIG. 2. Typical etch profile using Ni as the etch mask under base conditions. Vertical striations were transferred from imperfections in the mask edge.

FIG. 3. The etch rates of GaN, $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$, and AlN as a function of dc bias.

for GaN relative to $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$ and AlN over the range of $-(20-50\text{ V})$. The selectivity is the ratio of the etch rate of GaN to the AlN or $\text{Al}_x\text{Ga}_{1-x}\text{N}$. At -50 V , the selectivity between GaN and AlN was 8.5; whereas, it was only 1.2 between the GaN and $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$. The greatest selectivities for GaN were found at a bias of -20 V , a factor of 38 over AlN and approximately 10 over $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$. These differences in etch rates are consistent with the different bond energies between Ga–N and Al–N of 8.92 and 11.52 eV/atom, respectively.¹⁴ A second factor is the lower volatility of AlCl_x relative to GaCl_x . Since lower dc biases were used to attain the selective etching, there is a tradeoff between the selectivity and the total etch rate.

The strong dependence of the etch rate on dc bias indicates that ion bombardment plays a significant role in the

FIG. 4. The selectivity of GaN relative to $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$ and AlN as a function of dc bias. Values of selectivity were obtained from the GaN/AlN and $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$ etch rate ratios using the data for the individual etch rates shown in Fig. 3.

etching of these materials. Ion bombardment can enhance etching via damaging the surface to make it more reactive, stimulating desorption of the etch products, and direct physical sputtering. The existence of a threshold bias indicates that breaking Ga–N or Al–N bonds by ion bombardment may be the rate-limiting step. It is presumed that ion damage increases the reactivity of these ordinarily inert materials. The ion-induced damage may be necessary to form the volatile $\text{GaCl}_x/\text{AlCl}_x$ etch products.

In summary, dry etching of GaN, AlN, and $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$ have been investigated in an ICP system using Cl_2 and Ar as the process gases. The rates for all three materials depended strongly on both the ICP power and the dc bias. Maximum etch rates of 9800 Å/min for the GaN, 9060 Å/min for the $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$, and 7490 Å/min for the AlN were achieved which are the highest reported to date for this chemistry. Threshold biases of > -50 and -40 V were required to induce significant etching for AlN and $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$. No threshold was found for GaN down to -20 V . As a result, selectivities of 38 between GaN and AlN and 10 between GaN and $\text{Al}_{0.28}\text{Ga}_{0.72}\text{N}$ were obtained at a dc bias of -20 V . This is of potential interest for the fabrication of $\text{Al}_x\text{Ga}_{1-x}\text{N}$ based heterostructure devices. Research is ongoing to quantify the effects of plasma induced damage and to better understand the underlying mechanisms.

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¹R. F. Davis, Proc. IEEE **79**, 702 (1991).

²H. Morkoç, G. B. Gao, M. E. Lin, B. Suerdlov, and M. Burns, J. Appl. Phys. **76**, 1363 (1994).

³S. J. Pearton, C. R. Abernathy, F. Ren, J. R. Lothian, P. W. Wisk, and A. Katz, J. Vac. Sci. Technol. A **11**, 1772 (1993).

⁴R. J. Shul, S. P. Kilcoyne, M. Hagerott Crawford, J. E. Parmeter, C. B. Vartuli, C. R. Abernathy, and S. J. Pearton, Appl. Phys. Lett. **66**, 1761 (1995).

⁵M. E. Lin, Z. F. Fan, Z. Ma, L. H. Allen, and H. Morkoç, Appl. Phys. Lett. **64**, 887 (1994).

⁶G. F. McLane, T. Monahan, D. W. Eckart, S. J. Pearton, and C. R. Abernathy, J. Vac. Sci. Technol. A **14**, 1046 (1996).

⁷C. B. Vartuli, S. J. Pearton, J. W. Lee, J. Hong, T. D. Mackenzie, C. R. Abernathy, and R. J. Shul, Appl. Phys. Lett. **69**, 1426 (1996).

⁸R. J. Shul, G. B. McClellan, S. A. Casalnuovo, D. J. Rieger, S. J. Pearton, C. Constatine, C. Barratt, R. F. Karlicek, Jr., C. Tran, and M. Schurman, Appl. Phys. Lett. **69**, 1119 (1996).

⁹C. B. Vartuli, S. J. Pearton, J. W. Lee, A. Y. Polyakov, M. Shin, D. W. Greve, M. Skronowski, and R. J. Shul, Electrochem. Soc. Interface **144**, 2146 (1997).

¹⁰C. B. Vartuli, S. J. Pearton, J. D. Mackenzie, C. R. Abernathy, and R. J. Shul, Electrochem. Soc. Interface **143**, L246 (1996).

¹¹E. R. Lory, Solid State Technology (Nov.), 117 (1984).

¹²A. D. Hanser, C. A. Wolden, W. G. Perry, R. Therrien, and R. F. Davis (unpublished).

¹³M. D. Bremser, W. G. Perry, T. Zheleva, N. V. Edwards, O. H. Nam, N. Parikh, D. E. Aspnes, and R. F. Davis, MRS Internet J. Nitride Semicond. Res. **1**, 8 (1996).

¹⁴R. J. Shul, R. D. Briggs, S. J. Pearton, C. B. Vartuli, C. R. Abernathy, J. W. Lee, C. Constatine, and C. Barratt, Mater. Res. Soc. Symp. Proc. **449**, 969 (1996).